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C25C 7/08

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(56) Documents cited

GB 1417259

(58) Field of search

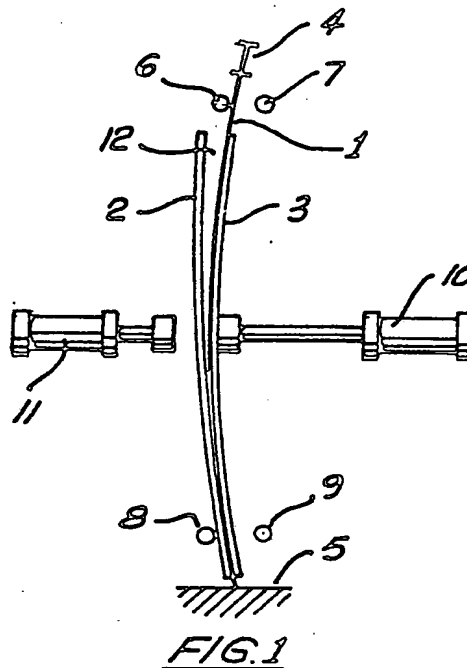
C7B

Selected US specifications from IPC sub-class C25C

(54) Stripping electrolytically deposited copper

(57) Electrolytically deposited copper is stripped from a cathode 1 by first flexing the cathode to cause at least part of the copper deposit to separate from the cathode, and then wedge-stripping or gas-blasting the remainder of the copper from the cathode.

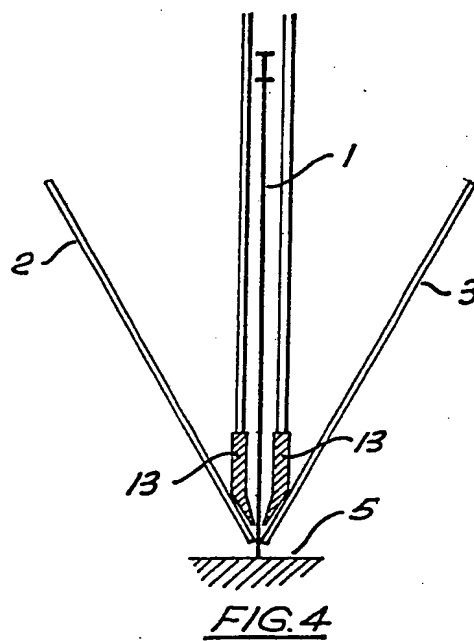
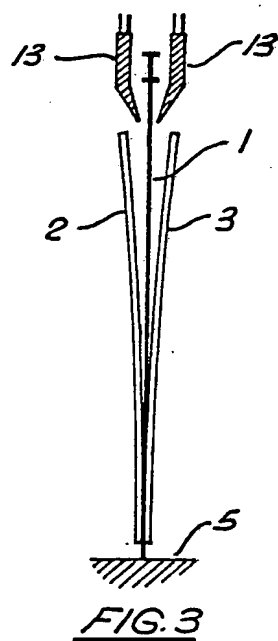
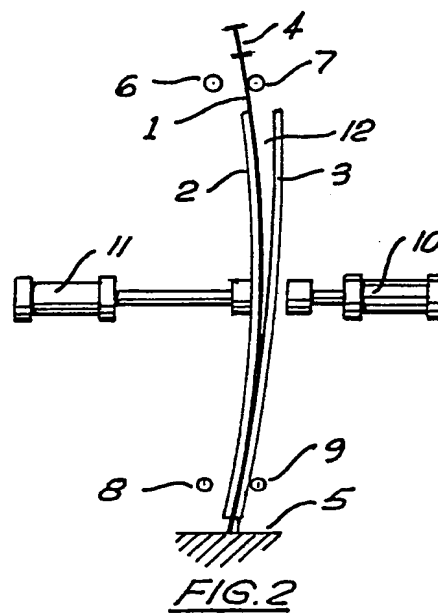
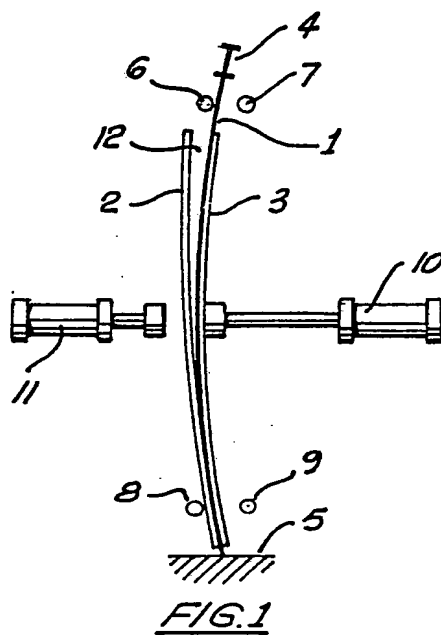
Preferably the cathode is a stainless steel sheet, and flexing is achieved by means of hydraulic rams 10,11 which press the centre of the sheet first in one direction and then in the opposite direction. The cathode is supported along its bottom edge (e.g. by surface 5) and held upright during the process (e.g. by reaction bars 6,7,8,9).



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SPECIFICATION

A method of stripping electrolytically deposited copper from a cathode

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FIELD OF INVENTION

This invention relates to a method of stripping electrolytically deposited copper from cathodes used in electrolytic cells.

- 10 The invention is applicable to any such cathodes but will be described with particular reference to the cathode described in Australian Patent No. 506,521. In that patent the cathode is a stainless steel starter sheet which
- 15 when suspended in the electrolytic bath and connected to a electric source accumulates a copper growth deposit on both sides.

DESCRIPTION OF PRIOR ART

- 20 In the prior art after the cathode is removed from the electrolytic bath the copper deposit on the sides of the upper end of the sheet is beaten with pneumatic hammers. This beating causes the copper deposit in that locality to
- 25 partially separate from the sheet. Then the remainder of the copper deposit is dislodged by the use of nozzled air blasts.

- A hammering action has the disadvantages of not only being noisy but also, over a period of time, distorting the stainless steel cathode sheet.
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SUMMARY OF INVENTION

- The invention consists in a method of stripping electrolytically deposited copper from a cathode comprising;
- 35 supporting said cathode along its lower edge,

- 40 flexing said cathode by an amount exceeding the strength of the adhesion bond between the deposited copper and the cathode but not exceeding the elastic limit of the cathode,

- 45 thereby causing at least part of the deposited copper to separate from the cathode and then wedge-stripping or gas-blasting the deposited copper from the cathode.

BRIEF DESCRIPTION OF DRAWINGS

- 50 The invention will now be described by way of example only with reference to the accompanying drawings in which:-

- Figure 1* is a view of one step in a flexing operation embodying the invention;

- 55 *Figure 2* is a view of a further step in that operation;

- Figure 3* is a view of a step in a wedge stripping operation embodying the invention; and

- 60 *Figure 4* is a view of a further stage of that wedge stripping operation.

DETAILED DESCRIPTION OF INVENTION

- 65 Electrolytic deposition of copper onto a stainless steel cathode sheet takes place while

the sheet is immersed in an electrolytic bath. The sheet being suspended from a hanger bar which hangs on electrically conductive rails above the electrolytic bath. The purpose of the electrolysis may be either:

- 70 electro-refining copper by collecting copper ions from impure copper anodes also suspended in the electrolytic bath; or, electro-winning copper ions from an electrolytic bath in which copper has been dissolved, and which contains inert or insoluble anodes, such as lead or lead alloys.

- 75 As the electrolysis proceeds a layer of copper is deposited on each side of the stainless steel cathode sheet. After a period of time the layer of copper increases to a thickness where it is desirable to remove it.

- Referring now to Fig. 1 the stainless steel cathode sheet 1 with its copper deposits 2 and 3 is removed from the electrolytic bath (not shown) and passed to a stripping station. Cathode 1 is not suspended from hanger bar 4 at the stripping station, but is supported on its bottom edge by supporting structure 5, and held substantially vertically between upper-pairs 6 and 7, and lower-pairs 8 and 9 of reaction bars. The cathode is preferably horizontally movable over a series of rollers on the surface of supporting structure 5. Reaction bars 6 and 7 hold an un-coated end of the cathode, while bars 8 and 9 hold a coated end.

- By avoiding suspending the sheet from hanger bar 4, during the stripping operation, abrasion of the copper clad electrically contacting ends is prevented. Since the bottom edge of the cathode sheet is recoated with wax before the sheet is re-used abrasion of the bottom edge is of no concern.

- 100 Preferably the middle portion of cathode 1 is pressed by one of a pair of hydraulic rams 10 and 11 until the cathode flexes. The head of the ram may make either point contact, line contact or area contact with the sheet. Hydraulic ram 10 flexes the centre of cathode 1 sufficiently to exceed the adhesion bond strength of copper deposit 2 which, as is shown, separates at least from the upper part of the stainless steel sheet at area 12.

- 105 Also preferably, as shown in Fig. 2, the other of the pair of hydraulic rams 11 then flexes cathode 1 in the opposite direction to displace the copper deposit 3 from the other side of the cathode.

- 120 It is also possible for separation of both deposits 2 and 3 to occur with flexing in only one direction.

- Alternatively, as cathode 1 is carried along, the upper and lower ends can be restrained in a series of rollers which as the cathode is moved horizontally will cause the cathode to flex when engaged by a series of rollers positioned in contact near the middle section of the cathode. It should be appreciated that many other methods of flexing a cathode are
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possible, and all fall within the scope of the invention.

- Once the edges of the upper portions of the copper deposits 2 and 3 are separated from cathode 1 a wedge 13, such as is shown in Fig. 3, is forced between each copper deposit 2 and 3 and the cathode 1 to completely separate the copper and the cathode. The copper then falls into a collecting bin. Fig. 4 depicts the final step of the wedge stripping with the wedge 13 at the bottom of cathode 1. Alternatively this can be accomplished by the use of air or other gas blasted from nozzles.
- Preferably the hydraulic cylinders on either side of the cathode are operated in such a manner as to be capable of deflecting the cathode over a range of 20 to 30mm in each direction by pushing at the centre. This is sufficient for the copper to be separated or sheared from the cathode down to the point where the hydraulic ram is pushing. The fact that this release at the top of the cathode is more positive than in the prior art allows an hydraulically operated wedge to be introduced behind the partially released copper growth to facilitate complete stripping. The average stripping rate can, as a consequence, be considerably speeded up. Currently the rate is an average of one sheet being stripped every six seconds.

As will be appreciated by those skilled in the art, the invention can be carried out when the cathode is held in any position other than vertically.

CLAIMS

1. A method of stripping electrolytically deposited copper from a cathode comprising:— flexing said cathode by an amount exceeding the strength of the adhesion bond between the deposited copper and the cathode but not exceeding the elastic limit of the cathode, thereby causing at least part of the deposited copper to separate from the cathode and then wedge-stripping or gas-blasting the deposited copper from the cathode.
2. A method according to Claim 1 in which the cathode is flexed in one direction only.
3. A method according to Claim 1 in which the cathode is flexed in one direction and then flexed in the opposite direction.
4. A method according to Claim 2 or 3 in which the cathode is flexed by a hydraulic ram which makes point, line or area contact with the cathode.
5. A method according to any preceding claim throughout which the cathode is held upright by two pairs of reaction bars, one pair on either side of the cathode near its lower edge and the other pair on either side of the cathode near its upper edge.
6. A method as claimed in any preceding claim in which the cathode is fabricated from

stainless steel.

7. A method according to Claim 6 in which the cathode is a sheet of stainless steel.

8. A method as hereinbefore described with reference to the accompanying drawings.

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